

# 醇气相氧化催化剂 Au/Ni-fiber 活性位 Ni<sub>2</sub>O<sub>3</sub>-Au<sup>+</sup>协同作用的光谱解释

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醇选择性催化氧化制备相应的醛酮是化工过程中非常重要的反应, 到目前为止, 许多催化剂被开发应用于该反应<sup>[1]</sup>。我们所开发的苯甲醇气相氧化制备苯甲醛催化剂Au/Ni-fiber, 具有有趣的活性结构NiO@Au (即较大的Au颗粒被较小的NiO碎片部分覆盖), 并且在Au颗粒和NiO界面处存在Ni<sub>2</sub>O<sub>3</sub>-Au<sup>+</sup>协同作用<sup>[2]</sup>。为了进一步研究该协同作用, 我们制备了以下三个样品: 在苯甲醇气流中通O<sub>2</sub>反应0.5 h制得工作状态下的样品1; 断O<sub>2</sub>后在苯甲醇气流中继续反应0.5 h制得样品2; 再次通O<sub>2</sub>反应0.5 h制得样品3。并采用XPS和XAFS对样品进行了谱学表征。结果表明(Fig. 1): 在O<sub>2</sub>通-断-通的过程中, 苯甲醇转化率和Au<sup>+</sup>含量都出现了高-低-高的剧烈变化, 伴随着这种剧烈变化, Ni<sub>2</sub>O<sub>3</sub>含量体现出轻微的高-低-高的变化, 但是NiO的变化规律恰好相反。由上可知: Au<sup>+</sup>和Ni<sub>2</sub>O<sub>3</sub>之间存在重要的协同关系, Au<sup>+</sup>是醇氧化的活性中心, 而Ni<sub>2</sub>O<sub>3</sub>即可以促使并稳定Au<sup>+</sup>的存在, 又可以起到活化O<sub>2</sub>的作用, 活性氧然后转移到Au<sup>+</sup>上实现对醇的氧化。

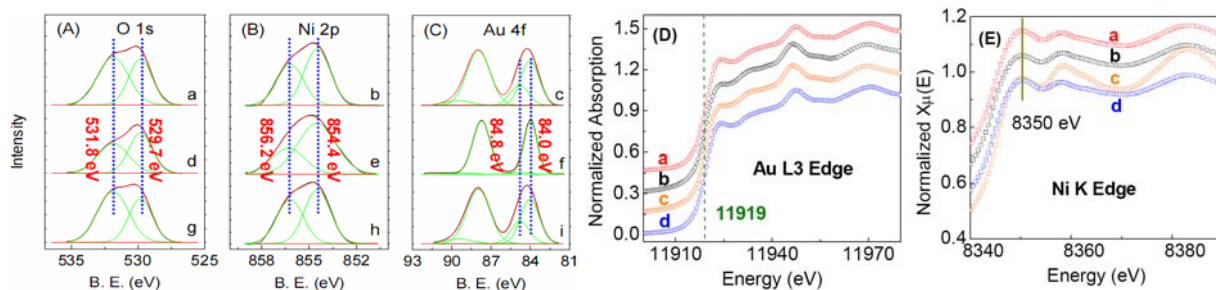


Fig. 1 XPS (A-C) and XANES (D-E) analysis of the catalyst Au/Ni-fiber over several samples. (A-C) a-c: sample 1; d-f: sample 2; g-i: sample 3; (D) a: sample 1; b: sample 2; c: Au foil; d: sample 3; (E) a: sample 1; b: sample 2; c: Ni foil; d: sample 3.

关键词: 金催化; 镍; 活性位; 有氧化; 醇

## 参考文献

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## The Spectroscopic Evidences for the Ni<sub>2</sub>O<sub>3</sub>-Au<sup>+</sup> Active Sites in Catalyst Au/Ni-fiber for the Low-temperature Gas-phase Oxidation of Alcohols

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An interesting date-cake-like ensembles of NiO@Au (*i.e.* 20-30 nm gold particle, like a cake, partially covered with very small NiO segments, like dates) are definitely identified to be highly active for low-temperature gas-phase oxidation of alcohols. On such NiO@Au ensembles, unique hybrid active-sites of Ni<sub>2</sub>O<sub>3</sub>-Au<sup>+</sup> are defined, bringing a step closer to the truth of the excellent low-temperature activity. By nature, the Ni<sub>2</sub>O<sub>3</sub> specimens not only promote the formation of Au<sup>+</sup> ions and stabilize them, but act as also an oxygen supplier that is proposed to transfer oxygen species onto the Au<sup>+</sup> sites to react with alcohol there.